Seat No.:	Enrolment No.

GUJARAT TECHNOLOGICAL UNIVERSITY

BE - IIIrd SEMESTER- EXAMINATION - WINTER 2014

Subject Code: 2132502 D Subject Name: Engineering Thermodynamics & Heat Transf			Date:18/12/ 2014 fer	
Tir	me: 2 tructio 1. 2.	2.30 p.m. – 5.00 p.m. Total Marks:	70	
Q.1	(a) (b)	What are different types of Thermodynamic Systems? Explain with the example. State the first law of thermodynamics. Mention its success and limitations?	07 07	
Q.2	(a) (b)	Derive general steady flow energy equation. Define following terms: state, path, process, perfect gas.	07 07 07	
	(b)	OR Prove that violation of Kelvin-Plank statement leads to violation of	07	
Q.3	(a)	Clausius statement. What is a cyclic process? Describe Carnot's cycle for establishing the maximum convertibility of heat into work?	07	
	(b)	Explain the concept of Energy & Exergy with the example. OR	07	
Q.3	(a) (b)	Explain Entropy. The change in entropy of a system, in a process is positive. Could the process be a spontaneous one? Explain. Write a short note on phase rule.	07 07	
Q.4	(a) (b)	Explain different modes of heat transfer along with their basic laws. Explain conduction and flow of heat through a composite wall when resistances are in series.	07 07	
Q.4	(a) (b)	OR What are various Types of Convection? Discuss natural convection in vertical and horizontal cylinders.	07 07	
Q.5	(a)	Derive the equation for logarithmic mean temperature difference (LMTD) for counter flow heat exchanger.	07	
	(b)	Draw the pool boiling curve and write the highlight of each segment in brief. OR	07	
Q.5	(a) (b)	Explain any two laws of Radiation. Derive the relation $A_1F_{12}=A_2F_{21}$ in relation to view factor.	07 07	
