Seat No.: Enroln	nent No.
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GUJARAT TECHNOLOGICAL UNIVERSITY B. Pharmacy Sem-II Examination June 2011

Subject code: 220006 Date: 02/07/11 **Subject Name: Physical Pharmacy-I** Time: 10:30am-1:30pm Total Marks: 80

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Instru	instructions:				
1.		pt any five questions.			
2.	Make	suitable assumptions wherever necessary.			
3.	Figure	s to the right indicate full marks.			
Q.1	(a)	Explain Phase equillibria for system containing two components.	06		
	(b)	Explain Liquid Crystalline State.	05		
	(c)	Explain Binding force between molecules.	05		
Q.2	(a)	Explain neutralization curves with necessary examples.	06		
	(b)	Explain Buffer Equation and Buffer Capacity.	05		
	(c)	Write Preparation of Pharmaceutical buffers and explain its	05		
		Applications.			
Q.3	(a)	Explain Solubility of Liquid in Liquid.	06		
	(b)	Give general principles of Solubility.	05		
	(c)	Explain Law of Distribution.	05		
Q.4	(a)	Write about the nernst and zeta potential and give its importance in	06		
		pharmaceutical systems.			
	(b)	What is spreading Coefficient? Derive its equation.	05		
	(c)	Write a note on Surface Active Agents.	05		
Q.5	(a)	Write a note on physical stability of emulsion.	06		
	(b)	Explain kinetic properties of Colloids.	05		
	(c)	Write a note on Controlled Flocculation.	05		
Q. 6	(a)	Discuss methods for determine Particle size and shape.	06		
	(b)	Name the two fundamental properties of powders. Explain the	05		
		application of derived properties of powders in Pharmacy.			
	(c)	Explain: Particle size and its distribution.	05		
Q.7	(a)	What a note on : Thixotropy and Negative Thixotropy	06		
	(b)	Explain Non Newtonian fluids with examples	05		

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How will you determine Rheological Properties?