

GUJARAT TECHNOLOGICAL UNIVERSITY
PDDC - SEMESTER – III • EXAMINATION – WINTER 2012

Subject code: X 31902**Date: 28/12/2012****Subject Name: Material Science and Metallurgy****Time: 10.30 am - 01.00 pm****Total Marks: 70****Instructions:**

1. Attempt all questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

- Q.1** (a) Classify engineering materials. Write any five Criteria for selection of materials for engineering applications. **07**
(b) Explain the various types of spark pattern to identify material in spark test. **07**
- Q.2** (a) Explain the difference between Ductile and Brittle structures. **07**
(b) Explain thermal equilibrium diagrams of binary alloy partial soluble in solid state. **07**
OR
(b) Explain phase and lever rule. **07**
- Q.3** (a) Draw iron carbon diagram and explain cooling process and its structure change at all critical points of hypo-eutectoid steel. **07**
(b) Explain the difference between mild steel and HSS. **07**
OR
- Q.3** (a) Draw iron carbon diagram and explain cooling process and its structure change at all critical points of hyper-eutectoid steel. **07**
(b) Explain structure, properties and application of malleable cast iron. **07**
- Q.4** (a) Describe effect of quenching media on properties of steel during heat treatment. **07**
(b) Explain TTT diagram. **07**
OR
- Q.4** (a) Mention all alloy of copper and explain any two. **07**
Q.4 (b) Explain cathodic protection against corrosion. **07**
- Q.5** (a) Explain working principle of LPT and its method and applications. **07**
(b) What is Powder metallurgy? State advantages and limitations of powder metallurgy. **07**
OR
- Q.5** (a) Explain working principle of MPT and its method and applications. **07**
(b) What is Wrought iron? Enlist properties and application of it. **07**
